

In Vivo Active Aldosterone Synthase Inhibitors with Improved Selectivity: Lead Optimization Providing a Series of Pyridine Substituted 3,4-Dihydro-1*H*-quinolin-2-one Derivatives

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Pyridine substituted naphthalenes (e.g., **I–III**) constitute a class of potent inhibitors of aldosterone synthase (CYP11B2). To overcome the unwanted inhibition of the hepatic enzyme CYP1A2, we aimed at reducing the number of aromatic carbons of these molecules because aromaticity has previously been identified to correlate positively with CYP1A2 inhibition. As hypothesized, inhibitors with a tetrahydronaphthalene type molecular scaffold (**1–11**) exhibit a decreased CYP1A2 inhibition. However, tetralone **9** turned out to be cytotoxic to the human cell line U-937 at higher concentrations. Consequent structural optimization culminated in the discovery of heteroaryl substituted 3,4-dihydro-1*H*-quinolin-2-ones (**12–26**), with **12**, a bioisostere of **9**, being nontoxic up to 200 μ M. The investigated molecules are highly selective toward both CYP1A2 and a wide range of other cytochrome P450 enzymes and show a good pharmacokinetic profile in vivo (e.g., **12** with a peroral bioavailability of 71%). Furthermore, isoquinoline derivative **21** proved to significantly reduce plasma aldosterone levels of ACTH stimulated rats.

Introduction

The progressive nature of congestive heart failure (CHF^a) is a consequence of a neurohormonal imbalance that involves a chronic activation of the renin–angiotensin–aldosterone system (RAAS) in response to reduced cardiac output and reduced renal perfusion. Aldosterone and angiotensin II (Ang II) are excessively released, leading to increased blood volume and blood pressure as a consequence of epithelial sodium retention as well as Ang II mediated vasoconstriction and finally to a further reduction of cardiac output.¹ The RAAS is pathophysiologically stimulated in a vicious cycle of neurohormonal activation that counteracts the normal negative feedback loop regulation. The most important circulating mineralocorticoid, aldosterone, acts by binding to specific mineralocorticoid receptors (MR) located in the cytosol of target epithelial cells. Thereby, renal sodium reabsorption and potassium secretion are promoted in the distal tubule and the collecting duct of the nephron. Elevated blood volume and thus blood pressure result from water that follows the sodium movement via osmosis. In addition to these indirect effects on the heart function, aldosterone exerts direct effects on the heart by activating nonepithelial MRs in cardiomyocytes, fibroblasts, and endothelial cells. Synthesis and deposition of fibrillar collagen in the fibroblasts result in myocardial fibrosis.² Relatively inelastic collagen fibers stiffen the heart muscle, which deteriorates the myocardial function and as a consequence enhances the neurohormonal imbalance by further stimulation of the RAAS. In addition to the effects of circulating aldosterone deriving from adrenal secretion, Satoh et al. reported that

aldosterone produced locally in the heart triggers myocardial fibrosis, too.³ Recent clinical studies with the MR antagonists spironolactone and eplerenone gave evidence for the pivotal role of aldosterone in the progression of cardiovascular diseases. Blocking the aldosterone action by functional antagonism of its receptor reduced the mortality and significantly reduced the symptoms of heart failure.⁴ Furthermore, follow-up studies revealed that cardiac fibrosis can not only be prevented but also reversed by use of spironolactone.⁵

However, several issues are unsolved by this therapeutic strategy. Spironolactone binds rather unselectively to the aldosterone receptor and also has some affinity to other steroid receptors, provoking adverse side effects.⁴ Although eplerenone is more selective, clinically relevant hyperkalemia remains a principal therapeutic risk.⁶ Another crucial point is the high concentration of circulating aldosterone, which is not lowered by MR antagonistic therapy and raises several issues. First, the elevated aldosterone plasma levels do not induce a homologous down-regulation but an up-regulation of the aldosterone receptor, which complicates a long-term therapy because MR antagonists are likely to become ineffective.⁷ Furthermore, the nongenomic actions of aldosterone are in general not blocked by receptor antagonists and can occur despite MR antagonistic treatment.⁸ A novel therapeutic strategy for the treatment of hyperaldosteronism, congestive heart failure, and myocardial fibrosis with potential to overcome the drawbacks of MR antagonists was recently suggested by us:^{9,10} blockade of aldosterone production by inhibiting the key enzyme of its biosynthesis, aldosterone synthase (CYP11B2), a mitochondrial cytochrome P450 enzyme that is localized mainly in the adrenal cortex and catalyzes the terminal three oxidation steps in the biogenesis of aldosterone in humans.¹¹ Consequent structural optimization of a hit discovered by compound library screening led to a series of nonsteroidal aldosterone synthase inhibitors with high selectivity versus other cytochrome P450 enzymes.^{12,13} Pyridine-substituted naphthalenes^{14,15} such as **I–III** (Chart 1) and dihydronaphthalenes,¹⁶ the most potent and selective compounds that emerged from our drug discovery program, however, revealed two major

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^a Abbreviations: ACTH, adrenocorticotrophic hormone; Ang II, angiotensin II; AUC, area under the curve; CHF, congestive heart failure; CYP, cytochrome P450; CYP11B1, 11*β*-hydroxylase; CYP11B2, aldosterone synthase; CYP17, 17*α*-hydroxylase-17,20-lyase; CYP19, aromatase; Het, heteroaryl; KHMDS, potassium hexamethyldisilazane; MR, mineralocorticoid receptor; NBS, *N*-bromosuccinimide; NCS, *N*-chlorosuccinimide; PPB, plasma protein binding; RAAS, renin–angiotensin–aldosterone system; Tf₂NPh, *N*-phenyltrifluoromethanesulphonimide.

Chart 1. Pyridynaphthalene Type CYP11B2 Inhibitors **I–III** and Design Strategy for 3,4-Dihydro-1*H*-quinolin-2-one Derivatives (e.g., **12** and **14**)

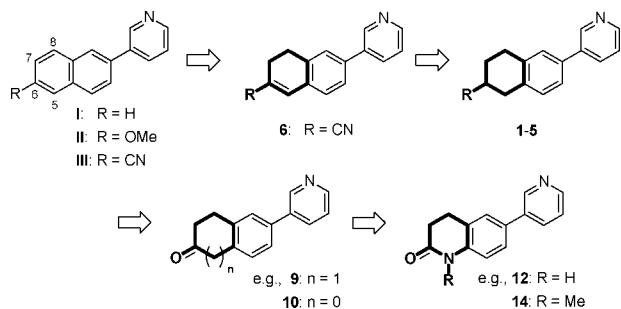
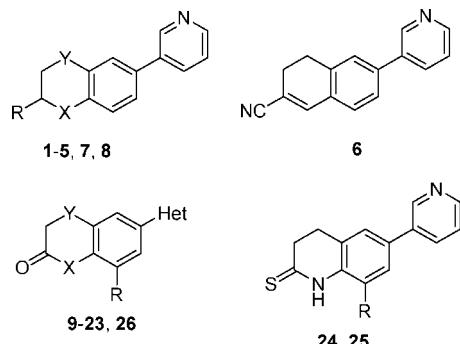


Chart 2. Title Compounds: Pyridine Substituted 3,4-Dihydro-1*H*-quinolin-2-one Derivatives and Structurally Related Compounds



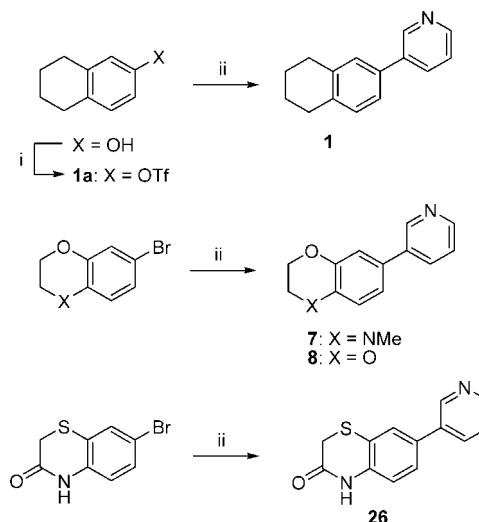
pharmacological drawbacks: a strong inhibition of the hepatic drug metabolizing enzyme CYP1A2 and no inhibitory effect on the aldosterone production *in vivo* by using a rat model.

In the present study, we describe the design and the synthesis of pyridine-substituted 3,4-dihydro-1*H*-quinolin-2-ones and structurally related compounds as highly potent and selective CYP11B2 inhibitors (Chart 2). The design concept toward these molecules is based on a systematic reduction of aromaticity by saturation of the hydrocarbons C₅ to C₈ of the naphthalene moiety and subsequent chemical modification of the fully saturated ring (Chart 1). The inhibitory activity of the title compounds was determined in V79 MZh cells expressing human CYP11B2. The selectivity was investigated with respect to the highly homologous 11*β*-hydroxylase (CYP11B1) as well as other crucial steroid- or drug-metabolizing cytochrome P450 enzymes (CYP17, CYP19, CYP1A2, CYP2B6, CYP2C9, CYP2C19, CYP2D6, and CYP3A4). The *in vivo* pharmacokinetic profile of some promising compounds was determined in male Wistar rats in both cassette and single dosing experiments. Furthermore, plasma protein binding and cytotoxicity studies were performed. The isoquinoline derivative **21** was investigated for aldosterone lowering effects *in vivo* using ACTH stimulated rats.

Results

Chemistry. The key step for the synthesis of the title compounds was a Suzuki coupling to introduce the heterocycle, mostly 3-pyridine. In case of the unsubstituted tetrahydronaphthalene **1**, the cross-coupling was accomplished by a microwave enhanced method¹⁷ using 3-pyridineboronic acid and triflate **1a**, which was prepared from the corresponding tetrahydronaphthalen-1-ol (Scheme 1).¹⁸ Compounds **7**, **8**, and **26** were synthesized via Suzuki coupling from commercially available arylbromides and 3-pyridineboronic acid under microwave heating.

Scheme 1^a

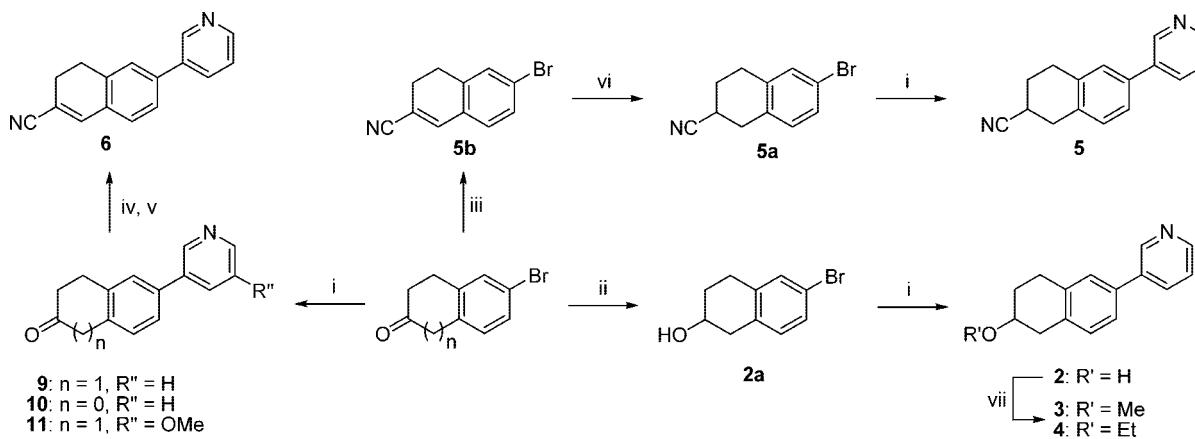
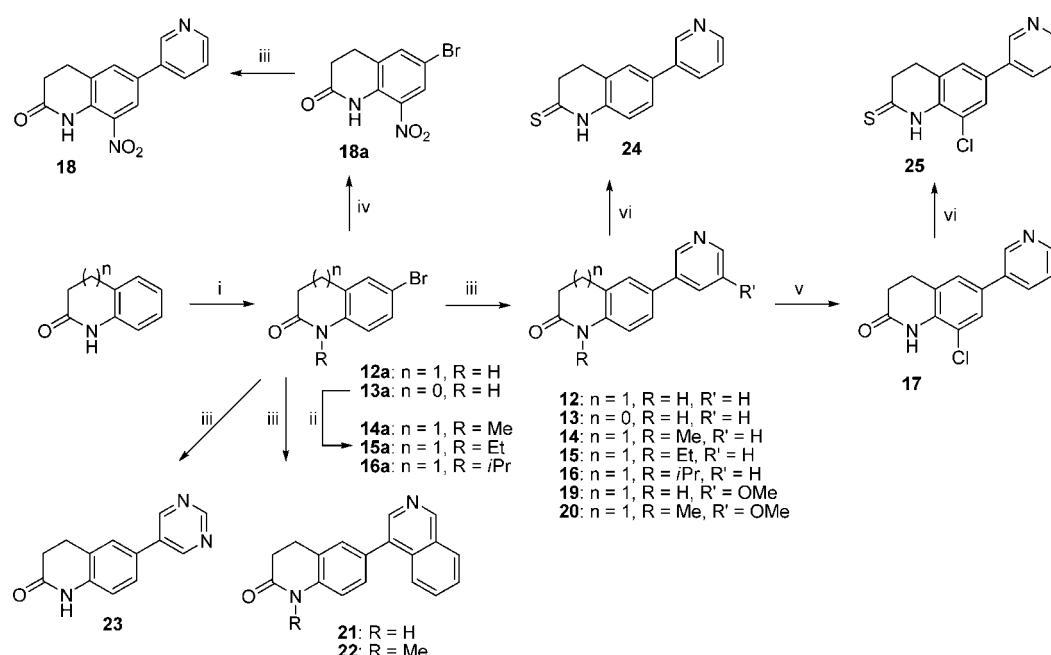


^a Reagents and conditions: (i) Tf₂NPh, K₂CO₃, THF, *μw*, 120 °C; (ii) 3-pyridineboronic acid, Pd(PPh₃)₄, aq NaHCO₃, DMF, *μw*, 150 °C.

The synthesis of compounds **2–6** and **9–11** (Scheme 2) started from either 6-bromo-2-tetralone (*n* = 1) or 5-bromo-1-indanone (*n* = 0) as key building blocks. Suzuki coupling afforded the heterocycle-substituted analogues **9–11**. Dihydroronaphthalene derivative **6** was prepared by treatment of tetralone **9** with KHMDS, *in situ* quenching the enolate with Tf₂NPh,¹⁹ and Pd-catalyzed cyanation²⁰ of the intermediate enoltriflate. The hydroxy-substituted tetrahydronaphthalene **2** was obtained by sodium borohydride reduction of the carbonyl group²¹ to afford **2a** and subsequent Suzuki coupling. *O*-Alkylation of **2** afforded the corresponding methoxy- (**3**) and ethoxy-substituted (**4**) derivatives. The 6-cyano-derivatized analogue **5** was prepared in three consecutive steps starting with a one-pot cyanohydrin/elimination step.²² The intermediate α,β -unsaturated nitrile **5b** was treated with NaBH₄ in refluxing ethanol to reduce the double bond.²³ Suzuki coupling of **5a** with 3-pyridineboronic acid afforded the tetrahydronaphthalene **5**.

The synthesis of the compounds with a dihydro-1*H*-quinolin-2-one and structurally related scaffold was accomplished as outlined in Scheme 3. The initial bromination procedures yielding either **12a** or **13a** have been described previously.^{24,25} Subsequent *N*-alkylation was accomplished by treating the quinolinones with alkyl halide and potassium *tert*-butylate in dimethylformamide to afford the intermediates **14a–16a**.²⁶ A nitro substituent was selectively introduced in the 8-position of **10a** by treating with a mixture of concentrated sulfuric acid and concentrated nitric acid to yield **18a**.²⁷ The obtained bromoarenes were transformed into the heterocycle-substituted analogues **12–16** and **18–23** by Suzuki coupling. Treating **12** with *N*-chlorosuccinimide in dimethylformamide afforded the 8-chloro derivative **17** as the only regioisomer. Conversion of the dihydroquinolinones **12** and **17** into the thio analogues **24** and **25** was carried out using Lawesson's reagent in refluxing toluene.

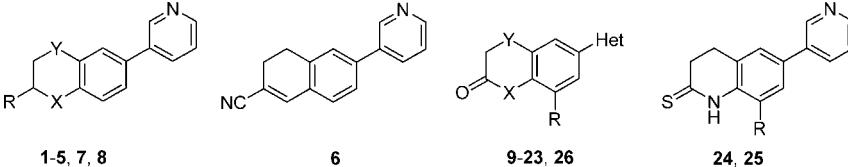
Biology. Inhibition of Human Adrenal Corticoid Producing CYP11B2 and CYP11B1 In Vitro (Table 1). The inhibitory activities of the compounds were determined in V79 MZh cells expressing either human CYP11B2 or CYP11B1.^{10,28} The V79 MZh cells were incubated with [¹⁴C]-deoxycorticosterone as substrate and the inhibitor in different concentrations. The product formation was monitored by HPTLC using a phosphoimager. Fadrozole, an aromatase (CYP19) inhibitor with

Scheme 2^aScheme 3^a

ability to reduce corticoid formation in vitro²⁹ and in vivo,³⁰ was used as a reference (CYP11B2, IC₅₀ = 1 nM; CYP11B1, IC₅₀ = 10 nM).

Most of the compounds presented in Table 1 show a strong inhibition of the target enzyme. Within the tetrahydro- and dihydronaphthalene series (compounds 1–6), a substituent in 6-position of the carbocyclic skeleton induces an increased inhibitory potency in case of methoxy- (compound 3) and cyano-substituents (compounds 5, 6) as well as a dramatic increase in selectivity toward CYP11B1, most notably in methoxy derivative 3 (selectivity factor = 347). Contrariwise, introduction of heteroatoms into the saturated ring leads to a decrease of CYP11B2 inhibition (e.g., 7 and 8). The carbonyl derivatives 9–11 are also highly potent (IC₅₀ = 1.8–7.8 nM) and selective aldosterone synthase inhibitors. Tetralone 9 is the most selective compound of the present series displaying a CYP11B1 IC₅₀ value 496-fold higher than the CYP11B2 IC₅₀ value. The derivatives with dihydro-1*H*-quinolin-2-one molecular scaffold

(12–26) exhibit a range of both inhibitory potency at the target enzyme (IC₅₀ = 0.1–64 nM) and selectivity with respect to CYP11B1 inhibition (selectivity factor = 44–440). A significant decrease in inhibitory potency can be observed for derivatization of the lactam nitrogen by an isopropyl residue (compound 16), whereas methyl (compound 14) and ethyl (compound 15) are tolerated in this position. The activity also decreases when the 3-pyridine is replaced by the 5-pyrimidine (compound 23) as the heterocyclic moiety. Introduction of a nitro substituent in the 8-position results in the rather moderate CYP11B2 inhibitor 18 (IC₅₀ = 64 nM) with lower CYP11B1 selectivity. Contrariwise, a chloro substituent in the same position (compound 17) increases the inhibitory potency by a factor of 7 compared to the hydrogen analogue 12. The most potent inhibitors are obtained when the 3-pyridine moiety is modified by 5-methoxylation or replaced by 4-isoquinoline, resulting in subnanomolar IC₅₀ values for compounds 20–22 (IC₅₀ = 0.1–0.2 nM). Isoquinoline derivative 22 displays an IC₅₀ value as low as 0.1

Table 1. Inhibition of Human Adrenal CYP11B2 and CYP11B1 In Vitro


compd	R	X	Y	Het	% inhibition ^a		IC ₅₀ value ^b (nM)		selectivity factor ^c
					V79 11B2 ^c hCYP11B2	V79 11B2 ^c hCYP11B2	V79 11B1 ^d hCYP11B1		
1	H	CH ₂	CH ₂		82	29	1977		68
2	OH	CH ₂	CH ₂		86	44	4921		112
3	OMe	CH ₂	CH ₂		97	3.3	1145		347
4	OEt	CH ₂	CH ₂		92	30	4371		146
5	CN	CH ₂	CH ₂		94	5.1	745		146
6					97	1.6	290		181
7	H	NMe	O		71	101	5970		59
8	H	O	O		70	154	13378		87
9	H	CH ₂	CH ₂	3-pyridine	97	7.8	3964		496
10	H	CH ₂	CH ₂	3-pyridine	90	4.4	819		186
11	H	CH ₂	CH ₂	5-methoxy-3-pyridine	94	1.8	191		106
12	H	NH	CH ₂	3-pyridine	88	28	6746		241
13	H	NH		3-pyridine	85	14	5952		425
14	H	NMe	CH ₂	3-pyridine	92	2.6	742		289
15	H	NEt	CH ₂	3-pyridine	93	22	5177		235
16	H	NiPr	CH ₂	3-pyridine	39	nd	nd		nd
17	Cl	NH	CH ₂	3-pyridine	97	3.8	1671		440
18	NO ₂	NH	CH ₂	3-pyridine	78	64	5402		84
19	H	NH	CH ₂	5-methoxy-3-pyridine	91	2.7	339		126
20	H	NMe	CH ₂	5-methoxy-3-pyridine	94	0.2	87		435
21	H	NH	CH ₂	4-isoquinoline	94	0.2	33		187
22	H	NMe	CH ₂	4-isoquinoline	99	0.1	6.9		69
23	H	NH	CH ₂	5-pyrimidine	57	nd	nd		nd
24	H				97	3.1	580		187
25	Cl				97	4.2	769		183
26	H	NH	S	3-pyridine	92	12	525		44
fadrozole						1.0	10		10

^a Mean value of at least four experiments, standard deviation less than 10%; inhibitor concentration, 500 nM. ^b Mean value of at least four experiments, standard deviation usually less than 25%, nd = not determined. ^c Hamster fibroblasts expressing human CYP11B2; substrate deoxycorticosterone, 100 nM. ^d Hamster fibroblasts expressing human CYP11B1; substrate deoxycorticosterone, 100 nM. ^e IC₅₀ CYP11B1/IC₅₀ CYP11B2, nd = not determined.

nM and is the most potent aldosterone synthase inhibitor known so far and also shows a pronounced inhibitory potency at CYP11B1 (IC₅₀ = 6.9 nM). The same trend was observed previously for the binding properties of a series of heteroaryl-substituted naphthalenes.³¹ Changing the lactam moiety into the corresponding thiolactam results in a slightly reduced CYP11B1 selectivity as seen in compounds **24** and **25** compared to the oxygen analogues **12** and **17**. Moreover, incorporation of a sulfur atom into the lactam moiety in compound **26** induces a dramatic loss of CYP11B1 selectivity (selectivity factor = 44).

Inhibition of Steroidogenic and Hepatic CYP Enzymes (Tables 2 and 3). The inhibition of CYP17 was investigated using the 50000g sediment of the *E. coli* homogenate recombinantly expressing human CYP17 and progesterone (25 μ M) as substrate.³² The percent inhibition values were measured at an inhibitor concentration of 2.5 μ M. Most of the investigated compounds display rather low inhibitory action on CYP17 (Table 2). However, a distinct inhibition in the range of 31–74% at a concentration of 2.5 μ M is observed in case of the tetrahydro- and dihydronaphthalenes **2–6**, which is comparable to the naphthalene parent compounds **I–III** (40–73%). All other derivatives, including the keto analogues **9–11** and the investigated dihydro-1*H*-quinolin-2-ones, are considerably less active at CYP17 (<22%). Exceptions from this are lactam **12** (41% inhibition) and the thiolactam analogue **24** (72%). The inhibition of CYP19 at a concentration of 500 nM was determined in vitro with human placental microsomes and [1β -³H]androstenedione as substrate as described by Thompson and Siiteri³³ using our

Table 2. Inhibition of Human CYP17, CYP19, and CYP1A2 In Vitro

compd	% inhibition ^a [IC ₅₀ value ^b (μ M)]		
	CYP17 ^c	CYP19 ^d	CYP1A2 ^e
I	40	[5.7]	99 [nd]
II	72	[0.6]	98 [nd]
III	73	[>36]	97 [nd]
2	49	23	74 [0.60]
3	44	22	80 [0.44]
4	31	10	73[0.62]
5	38	<5	72 [0.66]
6	74	<5	90 [0.18]
9	20	31	60 [1.55]
10	<5	53	57 [1.55]
11	21	58	57 [1.56]
12	41	21	50 [1.95]
13	<5	62	25 [6.58]
14	8	54	53 [1.79]
15	<5	63	36 [3.48]
17	7	17	14 [30.6]
19	12	<5	25 [5.24]
20	<5	5	20 [16.5]
21	22	<5	<5 [>150]
24	72	<5	77 [0.64]

^a Mean value of three experiments, standard deviation less than 10%.

^b Mean value of two experiments, standard deviation less than 5%. ^c *E. coli* expressing human CYP17; substrate progesterone, 25 μ M; inhibitor concentration, 2.5 μ M; ketoconazole, IC₅₀ = 2.78 μ M. ^d Human placental CYP19; substrate androstenedione, 500 nM; inhibitor concentration, 500 nM; fadrozole, IC₅₀ = 30 nM. ^e Recombinantly expressed enzymes from baculovirus-infected insect microsomes (Supersomes); inhibitor concentration, 2.0 μ M; furafylline, IC₅₀ = 2.42 μ M.

Table 3. Inhibition of Selected Hepatic CYP Enzymes In Vitro

compd	IC ₅₀ value ^a (μM)				
	CYP 2B6 ^{b,c}	CYP 2C9 ^{b,d}	CYP 2C19 ^{b,e}	CYP 2D6 ^{b,f}	CYP 3A4 ^{b,g}
3	47.4	16.3	<200	<200	12.2
9	<50	28.3	41.5	171	6.21
10	<50	123	147	<200	<200
11	<50	12.9	45.4	<200	3.8
12	<50	58.9	<200	171	127
14	<50	125	122	<200	<200
21	<100	2.9	9.2	<50	<50

^a Mean value of two experiments, standard deviation less than 5%.

^b Recombinantly expressed enzymes from baculovirus-infected insect microsomes (Supersomes). ^c Tranylcypromine, IC₅₀ = 6.24 μM. ^d Sulphaphenazole, IC₅₀ = 318 nM. ^e Tranylcypromine, IC₅₀ = 5.95 μM. ^f Quinidine, IC₅₀ = 14 nM. ^g Ketoconazole, IC₅₀ = 57 nM.

modification.³⁴ In most cases, the inhibitory action on CYP19 is low (<30%) at the chosen concentration (Table 2). Exceptions are observed in case of the keto derivatives **10** and **11** as well as the lactam derivatives **13–15**, displaying aromatase inhibition in the range of 53–63%.

A selectivity profile relating to inhibition of crucial hepatic CYP enzymes (CYP1A2, CYP2B6, CYP2C9, CYP2C19, CYP2D6, and CYP3A4) was determined by use of recombinantly expressed enzymes from baculovirus-infected insect microsomes. As previous studies have shown that aldosterone synthase inhibitors of the naphthalene type are potent inhibitors of CYP1A2 but otherwise rather selective versus important hepatic CYP enzymes,^{14,15} most of the newly prepared compounds were initially tested for their inhibitory action on CYP1A2 (Table 2). The parent compounds **1–III** are highly potent inhibitors of CYP1A2 (>95% inhibition at concentration of 2 μM). With regard to this high CYP1A2 activity of these naphthalene type compounds, the inhibitory potency slightly decreases in case of the dihydronaphthalene derivative **6** (90%) and the tetrahydronaphthalene derivatives **2–5** (72–80%). The keto analogues **9–11** exhibit 57–60% inhibition corresponding with IC₅₀ values of approximately 1.6 μM. A further decrease of CYP1A2 inhibition to less than 55% is observed for the investigated lactam bioisosteres **12–15**, **17**, and **19–21**. This is especially true for chloro-substituted **17** as well as compounds **20** and **21** with modified heterocycles displaying IC₅₀ values greater than 15 μM but also for indanone **13** and methoxypyridine derivative **19** with IC₅₀ values greater than 5 μM. However, compound **24**, the thio analogue of **12**, displays a pronounced inhibitory potency (IC₅₀ = 0.64 μM). Some compounds were also scrutinized for inhibition of other crucial hepatic CYP enzymes (Table 3). The data presented in Table 3 reveal that the investigated CYP enzymes are rather unaffected by the compounds with tetrahydronaphthalene (**3**), tetralone (**9–11**), as well as dihydro-1*H*-quinolin-2-one (**12**, **14**, and **21**) type molecular scaffold, and with few exceptions (i.e., **9**, **11** at CYP3A4 and **21** at CYP2C9) the IC₅₀ values measured are significantly greater than 10 μM.

Plasma Protein Binding (Table 4). The plasma protein binding of compounds **9**, **10**, and **12** was determined by ultrafiltration. Test solutions of an aliquot of concentrated test compound and rat or human plasma were incubated at 37 °C for 1 h and then centrifuged at 8000g for 20 min. Ultrafiltrates were analyzed for drug concentrations by LC-MS/MS. The plasma protein binding of the investigated CYP11B2 inhibitors was found to be low. The bound form of keto compounds **9** and **10** range between 22–25% in both human and rat plasma. In the case of the bioisosteric dihydro-1*H*-quinolin-2-one **12**,

Table 4. Plasma Protein Binding of Compounds **9**, **10**, and **12**

compd	PPB ^a (% bound)	
	rat	human
9	25	24
10	24	22
12	60	61

^a Determined by analysis of the ultrafiltrates via LC-MS/MS; the degree of binding to the plasma proteins (PPB) is calculated by the following equation: % PPB = (1 - [ligand_{ultrafiltrate}]/[ligand_{total}]) · 100.

Table 5. Pharmacokinetic Profile of Compounds **9–12**, **14**, and **21**

compd	dose (mg/kg) ^a	t _{1/2z} (h) ^b	t _{max} (h) ^c	C _{max} (ng/mL) ^d	AUC _{0–∞} (ng·h/mL) ^e
9^f	5	3.5	4	104	727
10^f	5	2.4	4	300	3178
11^f	5	3.9	4	35	212
12^{f,g}	5	3.8	4	261	1753
12	25	1.2	1	1537	4762
12	1 ^h	1.7			270
14^f	5	2.3	6	86	659
21	25	2.9	2	134	1658
fadrozole ^{f,g}	5	3.2	1	471	3207

^a Compounds were applied perorally to male Wistar rats. ^b Terminal half-life. ^c Time of maximal concentration. ^d Maximal concentration. ^e Area under the curve. ^f These compounds were investigated in a cassette dosing approach. ^g Mean value of two experiments. ^h Intravenous application.

the amount of freely available compound is lower (approximately 60% bound).

In Vivo Pharmacokinetics (Table 5). The pharmacokinetic profile of selected compounds was determined after peroral application to male Wistar rats. Plasma samples were collected over 24 h and plasma concentrations were determined by HPLC-MS/MS. Compounds **9–12** and **14** were investigated in cassette dosing experiments (peroral dose = 5 mg/kg) and compared to fadrozole. All five compounds show comparable absorption rates (t_{max} = 4–6 h) and terminal half-lives (t_{1/2z} = 2.3–3.9 h). The slowest elimination is observed in case of tetralone **11** (t_{1/2z} = 3.9 h) and dihydro-1*H*-quinolin-2-one **12** (t_{1/2z} = 3.8 h). Within this series, compound **10** shows the highest maximal concentration (C_{max}) in plasma (300 ng/mL), followed in the same range by compound **12** (261 ng/mL). Using the area under the curve (AUC_{0–∞}) as a ranking criterion, the bioavailability after peroral cassette dosing increases in the order **11** (212 ng·h/mL) < **14** (659 ng·h/mL) < **9** (727 ng·h/mL) < **12** (1753 ng·h/mL) < **10** (3178 ng·h/mL). Compounds **12** and **21** were investigated in single dosing experiments (peroral dose = 25 mg/kg). The amounts of the test compounds found in the plasma after peroral application are rather high in case of **21** (AUC_{0–∞} = 1658 ng·h/mL) and high in case of **12** (AUC_{0–∞} = 4762 ng·h/mL). Comparing the AUC of peroral with intravenous (dose = 1 mg/kg) application of dihydro-1*H*-quinolin-2-one **12** reveals an absolute bioavailability of 71%. Although the maximal concentration of **21** found in the plasma is significantly lower than in the case of compound **12** (factor 11), the AUC_{0–∞} values differ only by a factor of 3, which is due to the greater terminal half-life of isoquinoline derivative **21**. This becomes particularly apparent from Figure 1, where the mean profile of plasma levels (ng/mL) in rat versus time after oral application (25 mg/kg) of compounds **12** (Figure 1a) and **21** (Figure 1b) are shown. The concentrations of **21** are rather constant and measure between 90–130 ng/mL in a time frame of 0.5–8 h after application,

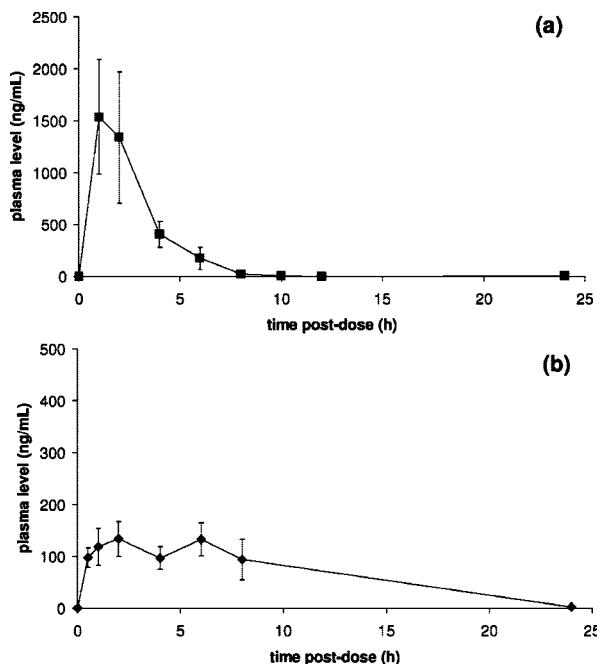


Figure 1. Mean profile (\pm SEM) of plasma levels (ng/mL) in rat versus time after oral application (25 mg/kg) of compounds **12** (a) and **21** (b) determined in single dosing experiments.

albeit the plasma levels are considerably lower than the plasma levels of **12**.

Discussion and Conclusion

Selectivity is a prerequisite of any drug candidate to avoid adverse side effects. In the development process of aldosterone synthase inhibitors, it is a crucial point to investigate the selectivity profile toward other cytochrome P450 enzymes at an early stage. It is known that the concept of heme-iron complexation (e.g., by nitrogen-containing heterocycles) is an appropriate strategy to discover highly potent and selective inhibitors. Because of this binding mechanism, however, a putative CYP11B2 inhibitor is potentially capable of interacting with other CYP enzymes by similarly binding to the heme cofactor with its metal binding moiety. Taking into consideration that the key enzyme of glucocorticoid biosynthesis, 11β -hydroxylase (CYP11B1), and CYP11B2 have a sequence homology of approximately 93%,³⁵ the selectivity issue is especially critical for the design of CYP11B2 inhibitors. Recently, we have demonstrated that 3-pyridine substituted naphthalenes such as **I–III** provide an ideal molecular scaffold for high inhibitory potency at the target enzyme CYP11B2 as well as high selectivity toward several other CYP enzymes (e.g., CYP11B1, CYP17, CYP19).¹⁴ However, these compounds strongly inhibit the hepatic enzyme CYP1A2 (e.g., compounds **I–III** in Table 2) that makes up about 10% of the overall cytochrome P450 content in the liver and metabolizes aromatic and heterocyclic amines as well as polycyclic aromatic hydrocarbons. In recent QSAR studies, CYP1A2 inhibition has been identified to correlate positively with aromaticity and lipophilicity.³⁶ Furthermore, both CYP1A2 substrates and inhibitors are usually small-volume molecules with a planar shape (e.g., caffeine³⁷ and furafylline³⁸). Rationalizing these findings, our design strategy aimed at reducing the aromaticity and disturbing the planarity of the molecules while keeping the pharmacophoric points³⁹ of the naphthalene molecular scaffold (see Chart 1).

These considerations led to the development of pyridine substituted dihydro- and tetrahydronaphthalenes **1–6**. The

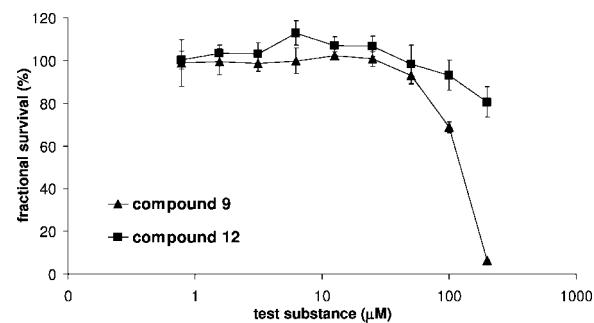


Figure 2. Mean profile (\pm SEM) of fractional survival (%) of human U-937 cells in presence of compounds **9** or **12**.

compounds are potent aldosterone synthase inhibitors with pronounced selectivity toward CYP11B1. As hypothesized, a decrease of CYP1A2 inhibition can be observed with decreased aromaticity (i.e., number of aromatic carbons) and planarity within this series. While the fully aromatized naphthalenes **I–III** exhibit 97–99% inhibition at a concentration of 2 μ M, dihydronaphthalene **6** is slightly less potent (90%) and the tetrahydro derivatives **1–5** are significantly less potent (72–80%) inhibitors of CYP1A2. However, IC₅₀ values are still below 1 μ M and thus the molecules are rather strongly inhibiting CYP1A2. Further increase in CYP1A2 selectivity is achieved by introduction of a keto group into the saturated ring as accomplished in compounds **9–11**. The inhibitory potencies toward CYP1A2 clearly decrease to IC₅₀ values of approximately 1.5 μ M. Presumably, the decrease in CYP1A2 inhibition is due to a reduced lipophilicity of the cyclic ketone scaffold compared to the tetrahydronaphthalene scaffold. Lipophilicity has been hypothesized to be one of the most important variables determining CYP1A2 inhibition.³⁶ Furthermore, the highly potent aldosterone synthase inhibitors **9** (IC₅₀ = 7.8 nM) and **10** (IC₅₀ = 4.4 nM) show reasonable plasma levels after peroral application to male Wistar rats (Table 5). However, due to the risk of the dialkyl ketone moiety to trigger unwanted cytotoxic effects, tetralone **9** was tested against the human cell line U-937 prior to further optimization. The experiment revealed that this compound significantly reduces the fractional survival of U-937 cells at concentrations greater than 100 μ M (Figure 2). Although the cytotoxic effect is seen only at rather high concentrations, which might not play a critical role under physiological conditions, a structural modification of the cyclic ketone moiety was performed with the aim to prevent this potential pharmacological drawback at an early stage by choosing a nontoxic scaffold for further structural modifications. Subsequent bioisosteric exchange by a lactam moiety gave rise to the dihydro-1*H*-quinolin-2-one derivatives **12–26**. Contrary to tetralone **9**, dihydro-1*H*-quinolin-2-one **12** exhibits no distinct cytotoxic effect on U-937 cells up to the highest concentration tested (Figure 2). In addition, compound **12** is an even slightly less potent inhibitor of CYP1A2 (IC₅₀ = 1.95 μ M) than the analogous tetralone (IC₅₀ = 1.55 μ M), which is again in correspondence with the reduced lipophilicity compared to the bioisosteric **9**. Contrariwise, lipophilicity does not influence the CYP1A2 potency within the series of dihydro-1*H*-quinolin-2-ones (**12–21**) as does the substitution pattern of the molecules, especially in the heterocyclic moiety (e.g., **20** and **21**). It is also striking that even minor structural variations such as introduction of a chloro substituent into the dihydro-1*H*-quinolin-2-one core as accomplished in **17** can induce an almost complete loss of CYP1A2 activity.

Pharmacokinetic investigations performed with compound **12** reveal a peroral absolute bioavailability of 71%. Compounds

14 and **21** are also capable of crossing the gastrointestinal tract and reach the general circulation after peroral application. However, their total range of absorption is below that of **12**. The plasma protein binding of inhibitor **12** was found to be low in both rat and human plasma (approximately 60%), indicating that a sizable free fraction of circulating compound is present in the plasma. Within the series of dihydro-1*H*-quinolin-2-one type inhibitors **12–26**, most compounds are highly active at the target enzyme. This particularly applies to the derivatives with a functionalized pyridine heterocycle, for example, methoxy derivative **20** ($IC_{50} = 0.2$ nM) and isoquinoline derivatives **21** ($IC_{50} = 0.2$ nM) and **22** ($IC_{50} = 0.1$ nM). By introduction of a chloro substituent in the 8-position or methoxy in the 5-position of the pyridine heterocycle, the selectivity increases to a factor of greater than 400 (**17**, **20**). Hence, these compounds are approximately 40-fold more selective than fadrozole (selectivity factor = 10). With respect to the high homology of the two CYP11B isoforms, this experimental result is particularly noteworthy.

To measure aldosterone-lowering effects in rats, we investigated the most potent and selective inhibitors of the present series for their ability to block aldosterone biosynthesis in V79 MZh cells expressing *rat* CYP11B2 prior to *in vivo* experiments. The results revealed that only compound **21** (and to a minor degree also the *N*-methyl analogue **22**) shows a moderate inhibitory action toward rat CYP11B2 *in vitro*, which corroborates our expectations concerning potential problems due to species cross-over, presumably as a consequence of a rather low sequence identity of the human and the rat CYP11B2 enzyme (approximately 69%).⁴⁰ The most potent compound in the rat assay, isoquinoline derivative **21**, is also a highly potent inhibitor of human CYP11B2 *in vitro* ($IC_{50} = 0.2$ nM) and exhibits a pronounced selectivity toward other CYP enzymes. Examination of availability in plasma following peroral administration of this compound to rats revealed a good half-life (2.9 h) and reasonable plasma levels ($AUC_{0-\infty} = 1658$ ng·h/mL following a 25 mg/kg dose). For the *in vivo* study, we used the animal model of Häusler et al., where the animals receive a subcutaneous injection of ACTH (1 mg/kg) 16 h before test item application to stimulate the gluco- and mineralocorticoid biosynthesis and the aldosterone levels are determined using RIA.^{40,41} The plasma aldosterone levels of the individual animals of this study are shown in Figure 3. It becomes apparent that ACTH treatment induces a significant increase of the aldosterone levels. Within the vehicle-treated group (animals 7–10, Figure 3b), the concentrations found in the plasma are rather constant over the duration of the experiment (6 h). To six animals (compound **21** group) isoquinoline derivative **21** was intravenously applied in a 20 mg/kg dose after 16 h upon ACTH treatment (animals 1–6, Figure 3a). A significant aldosterone-lowering effect can already be observed after 15 min. Within a time frame of 0.5–3 h, the aldosterone concentrations are maximally reduced to 36–63% of the prior ACTH level for the individual animals of the compound **21** group.

In summary, nonsteroidal aldosterone synthase inhibitors with a dihydro-1*H*-quinolin-2-one molecular scaffold are superior to the previously investigated pyridylnaphthalenes such as **I–III**. Most compounds exhibit a potent inhibitory activity at the target enzyme and isoquinoline derivative **22** is the most potent CYP11B2 inhibitor described so far ($IC_{50} = 0.1$ nM). The selectivity versus other steroidogenic as well as hepatic cytochrome P450 enzymes is generally high. Most notably, the strong inhibition of the hepatic CYP1A2 enzyme (>95% at a concentration of 2 μ M) present in the naphthalene type inhibitors

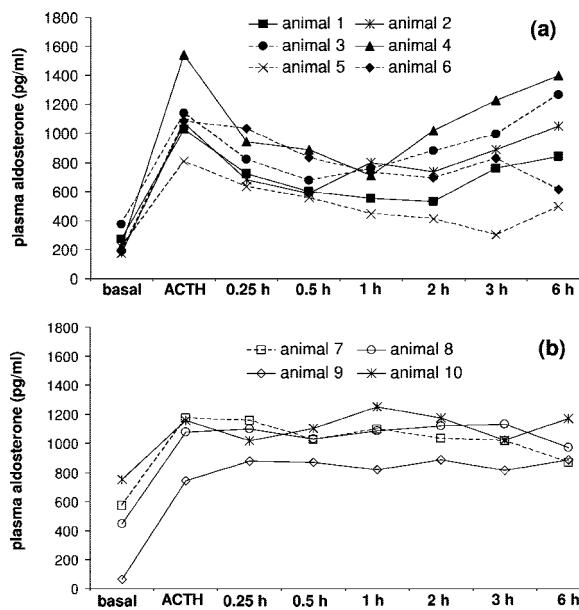


Figure 3. Lowering of aldosterone plasma levels *in vivo* of (a) the compound **21** treated group (animals 1–6) and (b) the vehicle treated group (animals 7–10). Shown are the aldosterone plasma levels of the individual animals predose before sc application of 1 mg/kg ACTH, and before iv application of 20 mg/kg **21** (animals 1–6) or vehicle (animals 7–10), followed by sampling 0.25, 0.5, 1, 2, 3, and 6 h postdose.

is significantly lower in the case of the dihydro-1*H*-quinolin-2-ones with IC_{50} values in the micromolar range. The investigated molecules reach the circulation after peroral administration to rats (e.g., **12** with a peroral bioavailability of 71%). Moreover, it has been found that isoquinoline derivative **21** significantly reduces the plasma aldosterone levels of ACTH stimulated rats. Our current research focuses on further *in vivo* investigations of compound **21** and structurally related compounds in disease oriented models to determine their capability to prevent or reverse myocardial fibrosis and reduce CHF induced mortality.

Experimental Section

Chemical and Analytical Methods. Melting points were measured on a Mettler FP1 melting point apparatus and are uncorrected. ^1H NMR and ^{13}C spectra were recorded on a Bruker DRX-500 instrument. Chemical shifts are given in parts per million (ppm), and tetramethylsilane (TMS) was used as internal standard for spectra obtained in DMSO-*d*₆ and CDCl₃. All coupling constants (*J*) are given in hertz. Mass spectra (LC/MS) were measured on a TSQ Quantum (Thermo Electron Corporation) instrument with a RP18 100-3 column (Macherey Nagel) and with water/acetonitrile mixtures as eluents. GC/MS spectra were measured on a GCD Series G1800A (Hewlett-Packard) instrument with an Optima-5-MS (0.25 μ M, 30 m) column (Macherey Nagel). Elemental analyses were carried out at the Department of Chemistry, University of Saarbrücken. Reagents were used as obtained from commercial suppliers without further purification. Solvents were distilled before use. Dry solvents were obtained by distillation from appropriate drying reagents and stored over molecular sieves. Flash chromatography was performed on silica gel 40 (35/40–63/70 μ M) with petroleum ether/ethyl acetate mixtures as eluents, and the reaction progress was determined by thin-layer chromatography analyses on Alugram SIL G/UV254 (Macherey Nagel). Visualization was accomplished with UV light and KMnO₄ solution. All microwave irradiation experiments were carried out in a CEM-Discover monomode microwave apparatus.

The following compounds were prepared according to previously described procedures: 6-bromo-1,2,3,4-tetrahydronaphthalen-2-ol

(2a),²¹ 6-bromo-3,4-dihydroquinolin-2(1*H*)-one (12a),²⁴ 5-bromo-1,3-dihydro-2*H*-indol-2-one (13a),²⁵ 6-bromo-8-nitro-3,4-dihydroquinolin-2(1*H*)-one (18a).²⁷

Synthesis of the Target Compounds. Procedure A.¹⁷ Boronic acid (0.75 mmol, 1 equiv), aryl bromide or -triflate (0.9–1.3 equiv), and tetrakis(triphenylphosphane)palladium(0) (43 mg, 37.5 μ mol, 5 mol %) were suspended in 1.5 mL DMF in a 10 mL septum-capped tube containing a stirring magnet. To this was added a solution of NaHCO₃ (189 mg, 2.25 mmol, 3 equiv) in 1.5 mL water and the vial was sealed with a Teflon cap. The mixture was irradiated with microwaves for 15 min at a temperature of 150 °C with an initial irradiation power of 100 W. After the reaction, the vial was cooled to 40 °C, the crude mixture was partitioned between ethyl acetate and water and the aqueous layer was extracted three times with ethyl acetate. The combined organic layers were dried over MgSO₄ and the solvents were removed in vacuo. The coupling products were obtained after flash chromatography on silica gel (petroleum ether/ethyl acetate mixtures) and/or crystallization.

Procedure B. Boronic acid (1 equivalent), aryl bromide or (1.3–1.5 equiv), and tetrakis(triphenylphosphane)palladium(0) (5 mol%) were suspended in toluene/ethanol 4/1 to give a 0.07–0.1 M solution of boronic acid under an atmosphere of nitrogen. To this was added a 1 N aqueous solution of Na₂CO₃ (6 equivalents). The mixture was then refluxed for 12–18 h, cooled to room temperature, diluted with water, and extracted several times with ethyl acetate. The combined extracts were dried over MgSO₄, concentrated, and purified by flash chromatography on silica gel (petroleum ether/ethyl acetate mixtures) and/or crystallization.

3-(5,6,7,8-Tetrahydronaphthalen-2-yl)pyridine (1). Compound **1** was obtained according to procedure A from **1a** (280 mg, 1.0 mmol) and 3-pyridineboronic acid (160 mg, 1.3 mmol) after flash chromatography on silica gel (petroleum ether/ethyl acetate, 2/1, R_f = 0.20) as a pale yellow oil (142 mg, 0.68 mmol, 68%), mp (HCl salt) 200–202 °C. LC/MS *m/z* 210.27 (MH⁺). Anal. (C₁₅H₁₅N·HCl) C, H, N.

6-Pyridin-3-yl-1,2,3,4-tetrahydronaphthalen-2-ol (2). Compound **2** was obtained according to procedure A from **2a** (114 mg, 0.50 mmol) and 3-pyridineboronic acid (80 mg, 0.65 mmol) after flash chromatography on silica gel (ethyl acetate, R_f = 0.27) as a colorless solid (96 mg, 0.43 mmol, 86%), mp 118–120 °C. LC/MS *m/z* 226.23 (MH⁺). Anal. (C₁₅H₁₅NO·0.1H₂O) C, H, N.

3-(6-Methoxy-5,6,7,8-tetrahydronaphthalen-2-yl)pyridine (3). To a suspension of NaH (73 mg, 1.84 mmol, 60% dispersion in oil) in 10 mL dry THF was added dropwise a solution of **2** (345 mg, 1.53 mmol) in 5 mL THF at room temperature. The mixture was heated to 50 °C until evolution of hydrogen ceased and then cooled to room temperature again. Thereupon, a solution of methyl iodide (326 mg, 2.30 mmol) in 5 mL THF was added via canula and stirring was continued at 50 °C for 3 h. The mixture was treated with saturated aqueous NH₄Cl solution and extracted three times with ethyl acetate. The combined organic extracts were washed with water and brine, dried over MgSO₄, and evaporated to dryness. The crude product was purified by flash chromatography on silica gel (petroleum ether/ethyl acetate, 7/3, R_f = 0.09) to afford **3** as a colorless oil (289 mg, 1.21 mmol, 79%), mp (HCl salt) 188–190 °C. LC/MS *m/z* 240.29 (MH⁺). Anal. (C₁₆H₁₇NO·HCl·0.6H₂O) C, H, N.

3-(6-Ethoxy-5,6,7,8-tetrahydronaphthalen-2-yl)pyridine (4). Compound **4** was obtained as described for **3** starting from **2** (270 mg, 1.20 mmol), NaH (58 mg, 1.44 mmol, 60% dispersion in oil), and ethyl bromide (196 mg, 1.80 mmol) after flash chromatography on silica gel (petroleum ether/ethyl acetate, 1/1, R_f = 0.31) as a colorless oil (198 mg, 0.78 mmol, 65%), mp (HCl salt) 186–188 °C. LC/MS *m/z* 254.29 (MH⁺). Anal. (C₁₇H₁₉NO·HCl·0.5H₂O) C, H, N.

6-Pyridin-3-yl-1,2,3,4-tetrahydronaphthalene-2-carbonitrile (5). Compound **5** was obtained according to procedure A from **5a** (130 mg, 0.55 mmol) and 3-pyridineboronic acid (88 mg, 0.72 mmol) after flash chromatography on silica gel (petroleum ether/ethyl acetate, 1/1, R_f = 0.17) as a colorless solid (91 mg, 0.39 mmol, 71%), mp 110–111 °C. ¹H NMR (500 MHz, CDCl₃): δ = 2.12 (m, 1H),

2.23 (m, 1H), 2.92 (m, 1H), 3.02–3.13 (m, 3H), 3.18 (dd, ²J = 16.4 Hz, ³J = 5.7 Hz, 1H), 7.19 (d, ³J = 7.9 Hz, 1H), 7.31 (s, 1H), 7.33–7.37 (m, 2H), 7.83 (m, 1H), 8.57 (dd, ³J = 5.0 Hz, ⁴J = 1.6 Hz, 1H), 8.80 (d, ⁴J = 1.6 Hz, 1H). ¹³C NMR (125 MHz, CDCl₃): δ 25.5, 26.1, 27.1, 32.1, 121.8, 123.5, 125.1, 127.8, 129.8, 132.3, 134.2, 135.5, 136.2, 136.4, 148.2, 148.5. LC/MS *m/z* 235.26 (MH⁺). Anal. (C₁₆H₁₄N₂·0.1H₂O) C, H, N.

6-Pyridin-3-yl-3,4-dihydronaphthalene-2-carbonitrile (6). To a solution of **6a** (562 mg, 1.58 mmol) in 10 mL degassed DMF were added zinc cyanide (117 mg, 1.00 mmol) and tetrakis(triphenylphosphane)palladium(0) (173 mg, 0.15 mmol) and the mixture was heated at 100 °C for 2 h. After cooling to room temperature, the mixture was diluted with 200 mL of water and extracted three times with ethyl acetate. The combined organic extracts were washed with water and brine, dried over MgSO₄, and evaporated to dryness. The crude product was crystallized from petroleum ether/ethyl acetate to afford **6** as colorless needles (286 mg, 1.23 mol, 78%), mp 142–143 °C. LC/MS *m/z* 233.23 (MH⁺). Anal. (C₁₆H₁₂N₂) C, H, N.

4-Methyl-7-pyridin-3-yl-3,4-dihydro-2*H*-1,4-benzoxazine (7). Compound **7** was obtained according to procedure A from 7-bromo-4-methyl-3,4-dihydro-2*H*-1,4-benzoxazine (228 mg, 1.00 mmol) and 3-pyridineboronic acid (160 mg, 1.30 mmol) after flash chromatography on silica gel (petroleum ether/ethyl acetate, 1/1, R_f = 0.26) as an off-white solid (102 mg, 0.45 mmol, 45%), mp 70–72 °C. LC/MS *m/z* 227.21 (MH⁺). Anal. (C₁₄H₁₄N₂O) C, H, N.

3-(2,3-Dihydro-1,4-benzodioxin-6-yl)pyridine (8). Compound **8** was obtained according to procedure A from 6-bromo-2,3-dihydro-1,4-benzodioxine (215 mg, 1.00 mmol) and 3-pyridineboronic acid (160 mg, 1.30 mmol) after flash chromatography on silica gel (petroleum ether/ethyl acetate, 7/3, R_f = 0.14) as a colorless solid (184 mg, 0.86 mmol, 86%), mp 59–61 °C. LC/MS *m/z* 214.19 (MH⁺). Anal. (C₁₆H₁₂N₂) C, H, N.

6-Pyridin-3-yl-3,4-dihydronaphthalen-2(1*H*)-one (9). Compound **9** was obtained according to procedure A from 6-bromo-2-tetralone (113 mg, 0.50 mmol) and 3-pyridineboronic acid (80 mg, 0.65 mmol) after flash chromatography on silica gel (petroleum ether/ethyl acetate, 1/1, R_f = 0.15) as a colorless oil (97 mg, 0.43 mmol, 86%), mp (HCl salt) 180–182 °C. LC/MS *m/z* 224.20 (MH⁺). Anal. (C₁₅H₁₃NO·HCl·0.4H₂O) C, H, N.

5-Pyridin-3-yl-2,3-dihydro-1*H*-inden-1-one (10). Compound **10** was obtained according to procedure A from 5-bromo-1-indanone (211 mg, 1.00 mmol) and 3-pyridineboronic acid (160 mg, 1.30 mmol) after flash chromatography on silica gel (petroleum ether/ethyl acetate, 1/1, R_f = 0.14) as a colorless solid (146 mg, 0.69 mmol, 69%), mp 122–123 °C. LC/MS *m/z* 210.69 (MH⁺). Anal. (C₁₄H₁₁-NO·0.1H₂O) C, H, N.

6-(5-Methoxypyridin-3-yl)-3,4-dihydronaphthalen-2(1*H*)-one (11). **11** was obtained according to procedure A from 6-bromo-2-tetralone (225 mg, 1.00 mmol) and 5-methoxy-3-pyridineboronic acid (199 mg, 1.30 mmol) after flash chromatography on silica gel (petroleum ether/ethyl acetate, 1/1, R_f = 0.15) as an off-white solid (133 mg, 0.53 mmol, 53%), mp 109–110 °C. LC/MS *m/z* 254.01 (MH⁺). Anal. (C₁₆H₁₅NO₂) C, H, N.

6-Pyridin-3-yl-3,4-dihydroquinolin-2(1*H*)-one (12). Compound **12** was obtained according to procedure B from **12a** (2.71 g, 12.0 mmol) and 3-pyridineboronic acid (1.23 g, 10.0 mmol) after crystallization from acetone/diethyl ether as colorless needles (2.15 g, 9.59 mmol, 96%), mp 181–183 °C. ¹H NMR (500 MHz, DMSO-*d*₆): δ 2.49 (t, ³J = 7.3 Hz, 2H), 2.95 (t, ³J = 7.3 Hz, 2H), 6.95 (d, ³J = 8.2 Hz, 1H), 7.43 (ddd, ³J = 7.9 Hz, ³J = 4.7 Hz, ⁵J = 0.6 Hz, 1H), 7.51 (dd, ³J = 8.2 Hz, ⁴J = 2.2 Hz, 1H), 7.56 (d, ⁴J = 2.1 Hz, 1H), 8.00 (ddd, ³J = 7.9 Hz, ⁴J = 2.2 Hz, ⁴J = 1.6 Hz, 1H), 8.50 (dd, ³J = 4.7 Hz, ⁴J = 1.5 Hz, 1H), 8.84 (d, ⁴J = 2.2 Hz, 1H), 10.19 (s, 1H). ¹³C NMR (125 MHz, DMSO-*d*₆): δ = 24.8, 30.3, 115.6, 123.8, 124.3, 125.6, 126.2, 130.6, 133.4, 135.2, 138.4, 147.2, 147.8, 170.2. LC/MS *m/z* 225.25 (MH⁺). Anal. (C₁₄H₁₂N₂O·0.1H₂O) C, H, N.

5-Pyridin-3-yl-1,3-dihydro-2*H*-indol-2-one (13). Compound **13** was obtained according to procedure A from **13a** (159 mg, 0.75 mmol) and 3-pyridineboronic acid (123 mg, 1.00 mmol) after crystallization

from acetone/diethyl ether as colorless needles (129 mg, 0.61 mmol, 81%), mp 218–220 °C. LC/MS *m/z* 211.01 (MH⁺). Anal. (C₁₃H₁₀N₂O·0.3H₂O) C, H, N.

1-Methyl-6-pyridin-3-yl-3,4-dihydroquinolin-2(1*H*)-one (14). Compound **14** was obtained according to procedure A from **14a** (110 mg, 0.46 mmol) and 3-pyridineboronic acid (74 mg, 0.60 mmol) after flash chromatography on silica gel (petroleum ether/ethyl acetate, 2/3, *R*_f = 0.07) as colorless needles (83 mg, 0.35 mmol, 76%), mp 100–101 °C. LC/MS *m/z* 239.80. Anal. (C₁₅H₁₄N₂O·0.1H₂O) C, H, N.

1-Ethyl-6-pyridin-3-yl-3,4-dihydroquinolin-2(1*H*)-one (15). Compound **15** was obtained according to procedure A from **15a** (229 mg, 0.90 mmol) and 3-pyridineboronic acid (92 mg, 0.75 mmol) after flash chromatography on silica gel (petroleum ether/ethyl acetate, 1/1, *R*_f = 0.09) and crystallization from acetone/diethyl ether as colorless plates (125 mg, 0.50 mmol, 55%), mp 91–92 °C. LC/MS *m/z* 253.00 (MH⁺). Anal. (C₁₆H₁₆N₂O·0.1H₂O) C, H, N.

1-(1-Methylethyl)-6-pyridin-3-yl-3,4-dihydroquinolin-2(1*H*)-one (16). Compound **16** was obtained according to procedure A from **16a** (174 mg, 0.65 mmol) and 3-pyridineboronic acid (74 mg, 0.60 mmol) after flash chromatography on silica gel (petroleum ether/ethyl acetate, 1/1, *R*_f = 0.14) as a colorless solid (47 mg, 0.18 mmol, 29%), mp 100–101 °C. LC/MS *m/z* 267.10 (MH⁺). Anal. (C₁₇H₁₈N₂O) C, H, N.

8-Chloro-6-pyridin-3-yl-3,4-dihydroquinolin-2(1*H*)-one (17). To a solution of **12** (560 mg, 2.50 mmol) in 5 mL DMF was added *N*-chlorosuccinimide (368 mg, 2.75 mmol) in 5 mL DMF over a period of 2 h at 65 °C. After additional 3 h at 65 °C, the mixture was poured into ice water and extracted three times with ethyl acetate. The combined organic layers were washed with water and brine, dried over MgSO₄, and the solvent was evaporated in vacuo. **17** was obtained after flash chromatography on silica gel (petroleum ether/ethyl acetate, 3/7, *R*_f = 0.15) and crystallization from acetone/diethyl ether as colorless needles (225 mg, 0.87 mmol, 35%), mp 177–178 °C. GC/MS *m/z* 258.95 (M⁺). Anal. (C₁₄H₁₁ClN₂O·0.1H₂O) C, H, N.

8-Nitro-6-pyridin-3-yl-3,4-dihydroquinolin-2(1*H*)-one (18). Compound **18** was obtained according to procedure B from **18a** (1.0 g, 3.70 mmol) and 3-pyridineboronic acid (546 mg, 4.44 mmol) after flash chromatography on silica gel (ethyl acetate, *R*_f = 0.15) as yellow needles (311 mg, 1.16 mmol, 31%), mp 187–189 °C. LC/MS *m/z* 269.94 (MH⁺). Anal. (C₁₄H₁₁N₃O₃) C, H, N.

6-(5-Methoxypyridin-3-yl)-3,4-dihydroquinolin-2(1*H*)-one (19). Compound **19** was obtained according to procedure A from **12a** (170 mg, 0.75 mmol) and 5-methoxy-3-pyridineboronic acid (150 mg, 0.98 mmol) after crystallization from acetone/diethyl ether as colorless needles (77 mg, 0.30 mmol, 40%), mp 213–215 °C. LC/MS *m/z* 255.02 (MH⁺). Anal. (C₁₅H₁₄N₂O₂) C, H, N.

6-(5-methoxypyridin-3-yl)-1-methyl-3,4-dihydroquinolin-2(1*H*)-one (20). Compound **20** was obtained according to procedure A from **14a** (200 mg, 0.83 mmol) and 5-methoxy-3-pyridineboronic acid (115 g, 0.75 mmol) after crystallization from acetone/diethyl ether as colorless needles (132 mg, 0.49 mmol, 66%), mp 158–159 °C. LC/MS *m/z* 268.95 (MH⁺). Anal. (C₁₆H₁₆N₂O₂·0.1H₂O) C, H, N.

6-Isoquinolin-4-yl-3,4-dihydroquinolin-2(1*H*)-one (21). Compound **21** was obtained according to procedure B from **12a** (1.55 g, 6.85 mmol) and 4-isoquinolineboronic acid (950 mg, 5.50 mmol) after crystallization from acetone/diethyl ether as colorless needles (800 mg, 2.92 mmol, 53%), mp 221–222 °C. LC/MS *m/z* 275.04 (MH⁺). Anal. (C₁₅H₁₄N₂O·0.1H₂O) C, H: calcd, 5.18, found, 5.60, N.

6-Isoquinolin-4-yl-1-methyl-3,4-dihydroquinolin-2(1*H*)-one (22). Compound **22** was obtained according to procedure A from **14a** (264 mg, 1.10 mmol) and 4-isoquinolineboronic acid (172 mg, 1.00 mmol) after crystallization from acetone/diethyl ether as colorless needles (163 mg, 0.57 mmol, 57%), mp 175–176 °C. LC/MS *m/z* 289.91 (MH⁺). Anal. (C₁₉H₁₆N₂O) C, H, N.

6-Pyrimidin-5-yl-3,4-dihydroquinolin-2(1*H*)-one (23). Compound **23** was obtained according to procedure A from **12a** (226 mg, 1.00 mmol) and 5-pyrimidineboronic acid (103 mg, 0.83 mmol) after

crystallization from ethanol as colorless needles (75 mg, 0.33 mmol, 40%), mp 232–233 °C. LC/MS *m/z* 225.74 (MH⁺). Anal. (C₁₃H₁₁N₃O·0.2H₂O) C, H, N.

6-Pyridin-3-yl-3,4-dihydroquinoline-2(1*H*)-thione (24). Compound **24** A suspension of **12** (395 mg, 1.76 mmol) and Lawesson's reagent (356 mg, 0.88 mmol) in dry toluene was refluxed for 30 min under an atmosphere of nitrogen. After cooling to room temperature, the solvent was removed in vacuo and the residue was purified by flash chromatography on silica gel (petroleum ether/ethyl acetate, 3/7, *R*_f = 0.31) to afford **24** as yellow plates (63 mg, 0.26 mmol, 15%), mp 265–267 °C. LC/MS *m/z* 241.05 (MH⁺). Anal. (C₁₄H₁₂N₂S) C, H, N.

8-Chloro-6-pyridin-3-yl-3,4-dihydroquinoline-2(1*H*)-thione (25). Compound **25** was obtained as described for **24** starting from **17** (900 mg, 3.48 mmol) and Lawesson's reagent (985 mg, 2.44 mmol) after flash chromatography on silica gel (ethyl acetate, *R*_f = 0.26) and crystallization from acetone/diethyl ether as yellow needles (212 mg, 0.77 mmol, 22%), mp 174–175 °C. GC/MS *m/z* 273.95 (M³⁵Cl⁺), 275.95 (M³⁷Cl⁺). Anal. (C₁₄H₁₁ClN₂S) C, H, N.

7-Pyridin-3-yl-2H-1,4-benzothiazin-3(4*H*)-one (26). Compound **26** was obtained according to general procedure B from 7-bromo-2H-1,4-benzothiazin-3(4*H*)-one (1.15 g, 4.71 mmol) and 3-pyridineboronic acid (695 mg, 5.56 mmol) after flash chromatography on silica gel (petroleum ether/ethyl acetate, 1/1, *R*_f = 0.20) and crystallization from ethanol as colorless needles (486 mg, 2.01 mmol, 43%), mp 238–240 °C. LC/MS *m/z* 242.99 (MH⁺). Anal. (C₁₃H₁₀N₂OS·0.2H₂O) C, H, N.

Biological Methods. **1. Enzyme Preparations.** CYP17 and CYP19 preparations were obtained by described methods: the 50000g sediment of *E. coli* expressing human CYP17³² and microsomes from human placenta for CYP19.³⁴

2. Enzyme Assays. The following enzyme assays were performed as previously described: CYP17³² and CYP19.³⁴

3. Activity and Selectivity Assay Using V79 Cells. V79 MZh 11B1 and V79 MZh 11B2 cells^{10,28} were incubated with [4-¹⁴C]-11-deoxycorticosterone as substrate and inhibitor in at least three different concentrations. The enzyme reactions were stopped by addition of ethyl acetate. After vigorous shaking and a centrifugation step (10000g, 2 min), the steroids were extracted into the organic phase, which was then separated. The conversion of the substrate was analyzed by HPTLC and a phosphoimaging system as described.

4. Inhibition of Human Hepatic CYP Enzymes. The recombinantly expressed enzymes from baculovirus-infected insect microsomes (Supersomes) were used and the manufacturer's instructions (www.gentest.com) were followed.

5. In Vivo Pharmacokinetics. Animal trials were conducted in accordance with institutional and international ethical guidelines for the use of laboratory animals. Cassette dosing: Male Wistar rats weighing 297–322 g (Janvier, France) were housed in a temperature-controlled room (20–24 °C) and maintained in a 12 h light/12 h dark cycle. Food and water were available ad libitum. The animals were anaesthetized with a ketamine (135 mg/kg)/xylazine (10 mg/kg) mixture and cannulated with silicone tubing via the right jugular vein. Prior to the first blood sampling, animals were connected to a counterbalanced system and tubing to perform blood sampling in the freely moving rat. Separate stock solutions (5 mg/mL) were prepared for the tested compounds in labrasol/water (1:1; v/v), leading to a clear solution. Immediately before application, the cassette dosing mixture was prepared by adding equal volumes of the stock solutions to end up with a final concentration of 1 mg/mL for each compound. The mixture was applied perorally to 3 rats with an injection volume of 5 mL/kg (Time 0). Blood samples (250 μL) were collected 1 h before application and 1, 2, 4, 6, 8, and 24 h thereafter. They were centrifuged at 650g for 10 min at 4 °C and then the plasma was harvested and kept at –20 °C until LC/MS analysis. To 50 μL of rat plasma sample and calibration standard 100 μL of acetonitrile containing the internal standard was added. Samples and standards were vigorously shaken and centrifuged for 10 min at 6000g and

20 °C. For the test items, an additional dilution was performed by mixing 50 µL of the particle free supernatant with 50 µL water. An aliquot was transferred to 200 µL sampler vials and subsequently subjected to LC-MS/MS. HPLC-MS/MS analysis and quantification of the samples was carried out on a Surveyor-HPLC-system coupled with a TSQ Quantum (ThermoFinnigan) triple quadrupole mass spectrometer equipped with an electrospray interface (ESI). The mean of absolute plasma concentrations (±SEM) was calculated for the three rats, and the regression was performed on group mean values. The pharmacokinetic analysis was performed using a noncompartment model (PK Solutions 2.0, Summit Research Services). Single dosing: the single dose experiments were performed as described for the cassette dosing procedure with male Wistar rats weighing 234–276 g (Janvier, France). Separate stock solutions (5 mg/mL) were prepared for compound **12** in PEG400/water/ethanol (50:40:10; v/v/v) and for compound **21** in labrasol/water (1:1; v/v), leading to clear solutions. Compound **12** was applied at 25 mg/kg perorally and 1 mg/kg intravenously and compound **21** at 25 mg/kg perorally to 4 rats each. Additional blood samples were taken 10 and 12 h after application in case of peroral application and 0.08, 0.25, 0.50, and 0.75 h in case of intravenous application of compound **12**, respectively.

6. Plasma Protein Binding. A 10 mM test compound solution and ketoprofen solution is prepared in acetonitrile. The test compound solution is diluted with solvent to the 50 fold concentration (150 µM, working solution) used in the assay (3 µM). In a 1.5 mL Eppendorf vial, 3 µL of working solution are given to 147 µL of serum (rat or human) and mixed. For recovery, the same dilutions are done in ultrafiltrated serum. The solutions are incubated for 1 h at 37 °C. The whole samples (6 test solutions, 6 recovery samples) are centrifuged at 8000g for 20 min using Centrifree micropartition devices (Millipore). The 75 µL of ultrafiltrate (UF) sample is removed for sample preparation. To 75 µL of sample or 150 µL calibration standard, 75 or 150 µL acetonitrile containing the internal standard (ketoprofen, 1 µM) is added to precipitate plasma proteins. Samples are then vigorously shaken (10 s) and centrifuged for 10 min at 6000g and 20 °C. An aliquot (70 µL) of the particle-free supernatant is subsequently subjected to LC-MS/MS. The degree of binding to the plasma proteins (PPB) is calculated by the following equation: % Protein binding = (1 – [ligand_{ultrafiltrate}]/[ligand_{total}]) · 100.

7. Cytotoxicity. Cell viability upon drug exposure was determined using a fluorimetric alamar blue conversion assay using a 96-well plate format. Briefly, U-937 cells (human monocytic leukemia) were seeded in growth medium into 96-well plates at a final density of 5×10^4 cells/mL and exposed to the respective compounds for the indicated time intervals (six replicates per concentration). At the end of the exposure time, alamar blue (Biosource International, Camarillo, CA) was added at 10% (v/v) and incubated for 4 h. Fluorescence intensity was quantitated using a Wallac Victor fluorescence plate reader (Perkin-Elmer, Wellesley, MA) at 530 nm excitation and 590 nm emission. Relative viability of cells was determined in relation to the untreated control. Control wells containing compound only were included to detect potential interference of the compound with the indicator system. Also, media only controls were included to account for background fluorescence. Viability of cells prior to cytotoxicity experiments was determined by Trypan Blue staining. Cells were diluted 1:3 in 0.4% (w/v) Trypan Blue (Sigma), and counted in a hemacytometer.

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Supporting Information Available: Aldosterone concentrations in the individual animals at all sampling points, NMR spectroscopic data of the target compounds **1–4**, **6–11**, **13–26**, full experimental details, and spectroscopic characterization of the reaction intermediates **1a**, **5a**, **5b**, **6a**, **14a–16a**, elemental analysis results of compounds **1–26**. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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